

Topic : Acid and Basic Strength
Type of Questions

Single choice Objective ('-1' negative marking) Q.1 to Q.6

(3 marks, 3 min.)

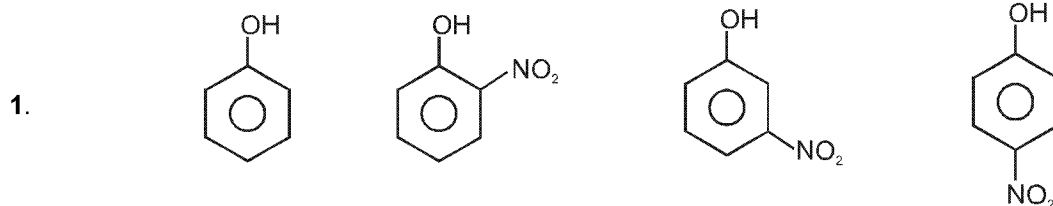
M.M., Min.

[18, 18]

Subjective Questions ('-1' negative marking) Q.7 to Q.8

(4 marks, 5 min.)

[8, 10]



	I	II	III	IV
Pka.	9.98	7.23	8.40	7.15

Mark True and False statements related to the variation in pKa values given for the compounds mentioned above

 S_1 : III is more acidic than I due to -I effect of $-\text{NO}_2$ group

 S_2 : Due to H-bonding in II, it is slightly less acidic than IV.

 S_3 : Due to presence of -M effect II & IV are more acidic than that of III.

 S_4 : Mesomeric effect of $-\text{NO}_2$ group is not operative in case of III

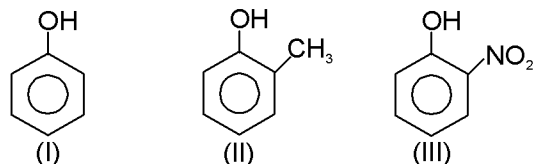
(A) TTTT

(B) TFTF

(C) FTFT

(D) FFTT

2. The correct decreasing order of acid strength of following compounds is :



(A) III > II > I

(B) I > II > III

(C) II > III > I

(D) III > I > II

3. Arrange basicity of the given compounds in decreasing order :

(i) $\text{CH}_3-\text{CH}_2-\text{NH}_2$ (ii) $\text{CH}_2=\text{CH}-\text{NH}_2$ (iii) $\text{CH}\equiv\text{C}-\text{NH}_2$

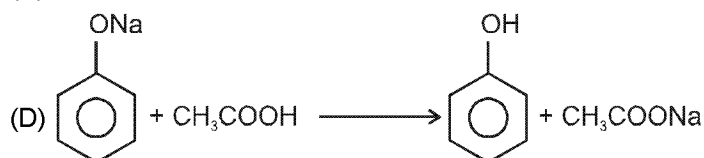
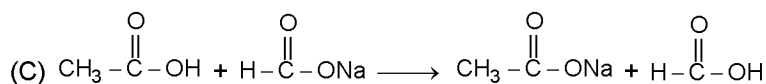
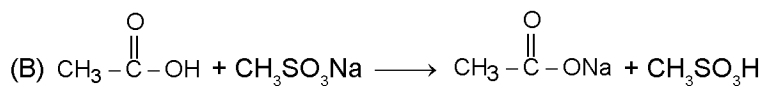
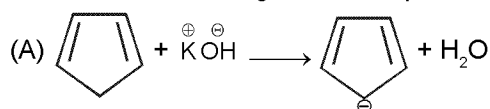
(A) i > ii > iii

(B) i > iii > ii

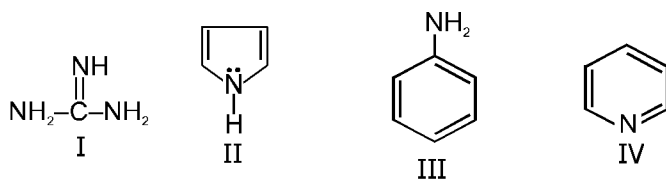
(C) iii > ii > i

(D) ii > iii > i

4. Which of the following reaction is possible ?

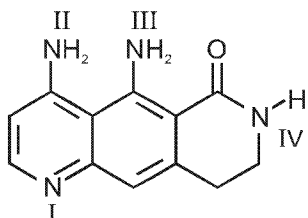


5. Write the basicity order of the following compounds?



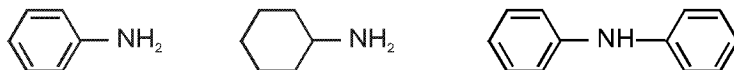
- (A) I > III > IV > II (B) I > II > III > IV (C) I > IV > III > II (D) I > III > II > IV

6. Identify the most basic nitrogen atom.

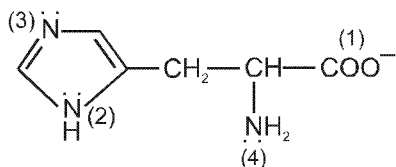


- (A) I (B) II (C) III (D) IV

7. Write the basicity order of the following compounds ?



8. Which of the following group is most basic :



Answer Key

DPP No. # 9

1. (A) 2. (D) 3. (A) 4. (D) 5. (C)
 6. (A) 7. II > I > III 8. 4

Hints & Solutions

DPP No. # 9

2. (D) $-\text{NO}_2 \rightarrow -M > -I$; $-\text{CH}_3 \rightarrow \text{H.C. and } +I$
In phenol, No ortho effect.

3. $-I$ effect \uparrow Basic strength \downarrow .